

# Synthesis and Antimicrobial activity of some Azaisatins derivatives containing 2-Phenylquinazolin-4(3H)-ones

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## Abstract:

Organic compounds having particular heterocyclic ring system have wide range of applications and used in pharmaceuticals, agrochemicals, veterinary products and for several other applications. Azaindoles have received considerable interest in view of their potential relationship to pharmacologically important indoles and purine nucleus, because of their interesting biological activities.

Hence a new azaisatins derivative containing 2-phenylquinazolin-4(3H)ones has been designed and synthesized. Their structures have been elucidated on the basis of elemental analysis and spectral studies. All the compounds synthesized were screened for their potential antimicrobial activities, which exhibited some authentic results towards testing organism invitro and invivo studies.

Keywords: Quinazoline, Azaisatin, Antibacterial, Antifungal.

## NTRODUCTION

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Quinazoline <sup>[1]</sup> is a versatile lead molecule for the design of potential bioactive agents because its derivative has been reported to possess broad spectrum biological activities. Quinazolinones are the oxidized form of guinazolines and as such are part the quinoline alkaloids. Several of quinazolinone alkaloids are known to elicit a wide including varietv of biological response antibacterial <sup>[2]</sup>, analgesic and anti-inflammatory <sup>[3]</sup>, antifungal <sup>[4]</sup> anticonvulsant <sup>[5]</sup>, anticancer <sup>[6]</sup>, and antihypertensive <sup>[7]</sup> activities.

It is evident from Literature that azaisatin <sup>[8]</sup> are also biologically active and found to have various pharmacological activities. So, keeping this in view the present work is to synthesize the title compounds to obtain derivatives of azaisatin.

## MATERIALS AND METHODS

Melting points were taken in open capillary tubes on Sigma-Aldrich melting point apparatus and are uncorrected. All the synthesized compounds were purified by thin layer chromatogram on silica gel G using toluene: ethyl acetate (7:3) and visualized with UV light. IR spectra were recorded on PERKIN-ELMER BX Series FTIR spectrometer using KBr pellets. <sup>1</sup>HNMR spectra were recorded in a CDCl<sub>3</sub> as a solvent and tetra methyl silane (TMS) as an interval standard. MASS Spectra of the compounds were recorded on a Agilent Mass spectroscopy 1100 series using ESI technique.

The physical constants of different synthesized azaisatins derivatives containing 2phenylquinazolin4-(3H)-ones are shown in Table No.1

The	spectral	data	of	different	synthe	esized
azais	atins	derivati	ves	conta	ining	2-

2-Aminobenzoic acid

I

H<sub>3</sub>C-

-он Methanol

R



No.2

3-(1,2-Dihydro-1-substituted-2-oxopyrrolo[2,3-b]pyridin-3-ylideneamino)-2-phenylquinazolin-4(3H)ones

IV (a-d)

Figure 1

#### GENERAL PROCEDURE

## A) Synthesis of 2-Phenyl-(4H)-3, 1-benzoxazin-4one [9] (II)

To a stirred solution of 2-aminobenzoic acid (I) (0.01 mole) in pyridine (6ml), benzoyl chloride (0.01 mole) was added drop wise maintaining the temperature near 8 0C for about 1.5 hours. Reaction mixture was stirred for another 3 hours at room temperature, while stirring a solid product separated out. Whole reaction mixture was neutralized with 5% NaHCO3 solution. A pale

yellow deposited product was filtered, washed with cold water and dried. The solid product was recrystallized from ethanol.

Yield: 86%, m.p: 114 – 116 ° C

#### B) Synthesis of 3-Amino-2-phenylquinazolin-4(3H)-one [9] (III)

In 100ml of round bottomed flask, a solution of hydrazine hydrate (0.10 mole) and 2-phenyl-4Hbenzo[d] [3, 1] oxazin-4-one (II, 0.05 mole) in methanol were taken and refluxed for 3-6 hours and the reaction was monitored by TLC. The

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solution was cooled, poured into cold water and the product was filtered, washed with cold water and dried. The solid product was recrystallized from ethanol.

Yield: 82%, m.p: 168-172 ° C

C) Synthesis of 3-(1, 2-Dihydro-1-substituted-2oxopyrrolo [2, 3-b] pyridin-3-ylideneamino)-2phenylquinazolin-4(3H)-ones [10] (IV) a-d

A mixture of 3-Phenyl-2-methylquinazolin-4(3H)one (VII, 0.001 mole) and substituted azaisatin (0.01 mole) in 10ml of glacial acetic acid were refluxed for 10-15minutes at 140 watt in Catalyst Systems Scientific microwave System and the reaction was monitored by TLC for completion. The resultant solution was poured into cold water. The product was filtered, washed with cold water and dried. The solid product was recrystallized from absolute ethanol.

 Table 1: PHYSICAL CONSTANTS OF DIFFERENT AZAISATINS DERIVATIVES CONTAINING

 2-PHENYLQUINAZOLIN-4-(3H)-ONES

Compound	R	Molecular weight (amu)	Molecular Formula	M.P	%yield	<b>Recrystallization solvent</b>
IV a	$-CH_3$	381	C <sub>22</sub> H <sub>15</sub> N <sub>5</sub> O <sub>2</sub>	216 <sup>0</sup> C-218 <sup>0</sup> C	72	Absolute Ethanol
IV b	$-C_{2}H_{5}$	395	C <sub>23</sub> H <sub>17</sub> N <sub>5</sub> O <sub>2</sub>	204 <sup>°</sup> C-206 <sup>°</sup> C	75	Absolute Ethanol
IV c	СН <sub>3</sub> СН <sub>3</sub>	409	C <sub>24</sub> H <sub>19</sub> N <sub>5</sub> O <sub>2</sub>	213ºC-215ºC	74	Absolute Ethanol
IV d	- C <sub>6</sub> H <sub>13</sub>	451	C <sub>27</sub> H <sub>25</sub> N <sub>5</sub> O <sub>2</sub>	208 <sup>0</sup> C-210 <sup>0</sup> C	73	Absolute Ethanol

 Table 2: Spectral Data

COMPOUND	IR(KBr)cm <sup>-1</sup>	¹H NMR(400 MHz, CDCl₃)	MASS SPECTRUM m/z
Synthesis of 3-Amino-2- phenylquinazolin-4(3H)-one <b>(III)</b>	3568.18,3309.03(d,NH,stretch), 1662.94 (C=N, stretch), 1718.24(C = O, stretch)	δ [ppm]: 8.306 (d, 1H, Ar-H), 7.800 (m,4H,Ar-H), 7.520 (m, 4H,Ar-H), 5.015 (s, 2H,NH <sub>2</sub>	The molecular ion was observed at 238 [M + H] <sup>+</sup> 260 [M +Na] <sup>+</sup> and 497 [2M +Na] <sup>+</sup>
3-(1, 2-Dihydro-1-ethyl-2- oxopyrrolo [2, 3-b] pyridin-3- ylideneamino)-2- phenylquinazolin-4(3H)-one. [ <b>IV</b> <b>b</b> ]	1718.17 (C = O, stretch), 1699.75 (C=O, stretch), 1654.24 (C=N, stretch	δ [ppm]: 8.334 (d, 1H, Ar-H), 8.095 (d, 1H,Ar-H), 7.104 (t, 1H,Ar-H), 7.844 (d,3H,Ar-H), 7.670 (m, 3H,Ar-H), 7.542 (m,3H,Ar-H), 3.987 (q, 2H, CH <sub>2</sub> ), 1.393 (t, 3H,CH <sub>3</sub> )	The molecular ion was observed at 396 [M + H] <sup>+</sup> and 418 [M +Na] <sup>+</sup>
3-(1, 2-Dihydro-1-isopropyl-2- oxopyrrolo [2, 3-b] pyridin-3- ylideneamino)-2- phenylquinazolin-4(3H)-one. <b>[IV</b> <b>c]</b>	1718.20 (C = O, stretch), 1677.36 (C=O, stretch), 1654.24 (C=N, stretch	δ [ppm]: 8.305 (d, 1H, Ar-H), 8.095 (d, 1H,Ar-H), 7.078 (t, 1H,Ar-H), 7.843 (d,3H,Ar-H), 7.669 (m, 3H,Ar-H), 7.540 (m,3H,Ar-H), 4.839 (m, 1H, CH), 1.615 (d, 6H,CH <sub>3</sub> )	The molecular ion was observed at 410 [M + H] <sup>+</sup> and 432 [M +Na] <sup>+</sup>

# RESULTS

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#### **Antibacterial Activity**

Antibacterial activity of compounds IV a-d was performed by disc diffusion method using nutrient agar medium against gram positive bacteria viz., Bacillus subtilis, Staphylococcus aureus, Streptococcus pneumonia and gram negative bacteria viz., E.coli, Proteus vulgaris, Klebsiella pneumonia, Pseudomonas aeruginosa at

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10µg/ml concentration. Ciprofloxacin was used as standard for comparison. The results are given in Table No.3

### Antifungal Activity

The compounds IV a-d were tested for their antifungal activity against four test organism Aspergillus niger, Aspergillus Flavus, Candida albicans, Fusarium oxysporium using potatodextrose-agar medium at 10µg/ml concentration. Inhibition zones were measured and the diameter was calculated in millimetre. The antifungal of these compounds was compared with standard drug Fluconazole and they have a promising activity. The results are given in Table No.4

## DISCUSSION

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## Antibacterial activity

Table No.3 shows the anti-bacterial activity data of 3-(1-substituted-1, 2-dihydro-2-oxopyrrolo-[2, 3b] pyridin-3-ylideneamino)-2-phenyl-quinazolin-4(3H)-ones.

Amongst them, compounds IV c, IV d have been found to be relatively more active against B.subtilis with a zone of inhibition of 18mm, and 19mm respectively, whereas compound IV d were effective against S.aureus with the zone of inhibition of 20mm respectively.

Compounds IV c, IV d were moderately effective against S. pneumonia with a zone of inhibition of 18mm and 17mm respectively, whereas compounds IV c, IV d were effective against E.coli with a zone of inhibition of 18mm, 19mm respectively.

Compounds IV b, IV c, IV d were more effective against P.vulgaris. with a zone of inhibition of 18mm, 20mm, 19mm respectively, whereas compounds IV b, IV d are effective against

K.pneumonia with a zone of inhibition of 20mm and 20mm respectively. Compound IV d was effective against P. aeruginosa with a zone of inhibition of 18mm.

### Antifungal activity

Table No.4 shows the antifungal activity data of 3-(1-substituted-1, 2-dihydro-2-oxopyrrolo-[2, 3-b] pyridin-3-ylideneamino)-2-phenyl-quinazolin-4(3H)-ones.

Amongst them, compounds IV c has been found to be effective against Aspergillus Niger with the zone of inhibition of 20mm. Compounds IV d have been found to be effective against Aspergillus Flavus with the zone of inhibition of 20mm. Compounds IV b, IV c, IV d have been found to be more effective against Candida albicans with the zone of inhibition of 19mm, 20mm, 19mm respectively, whereas compounds IV d have been found to be effective against Fusarium oxysporium with the zone of inhibition of 18mm.

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## Antibacterial activity

All the synthesized compound were found to be very active at a concentration of (10µg/disc) against the gram-positive and gram-negative organisms.

The compounds with isopropyl and hexyl substitutions at position 1 in azaisatin moiety were found to have good antibacterial activity against all the organisms used.

Compared to other three compounds IV d has been found to be relatively more effective against bacterial strains.

#### Antifungal activity

All the synthesized compound were found to be very active at a concentration of (10µg/disc) against fungal strains used.

The compounds with ethyl, isopropyl and hexyl substitutions at position 1 in azaisatin moiety were

found to have good antifungal activity against all

the organisms used.

Compounds IV c and IV d are found to be more effective against fungal strains.

Table 3: Antibacterial Activities of synthesized compounds [Zone of inhibition zone in mm at 10µg/disc]

Compounds	10µg/ml	B.subtilis	S.aureus	S. pneumonia	E.coli	P.vulgaris	K.pneumonia	P. aeruginosa
IV a	10	15	16	10	15	14	15	10
IV b	10	17	14	16	15	18	20	16
IV c	10	18	15	18	18	20	17	16
IVd	10	19	20	17	19	19	20	18
Ciprofloxacin	10	29	28	22	28	24	28	26

Table 4: Antifungal activities of synthesized compounds [Zone of inhibition zone in mm at 10µg/disc]

Compounds	10µg/ml	Aspergillus niger	Aspergillus Flavus	Candida albicans	Fusarium oxysporium
IV a	10	11	15	14	12
IV b	10	14	13	19	15
IV c	10	20	14	20	17
IVd	10	17	20	19	18
Fluconazole	10	27	26	29	24

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## CONFLICT OF INTEREST STATEMENT

We declare that we have no conflict of interest.

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